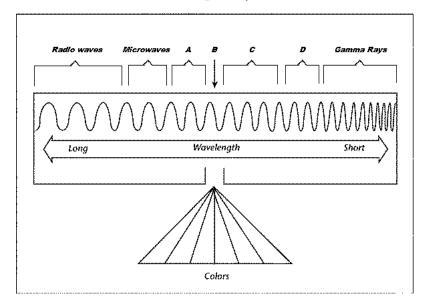
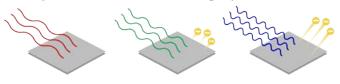
Electromagnetic Spectrum

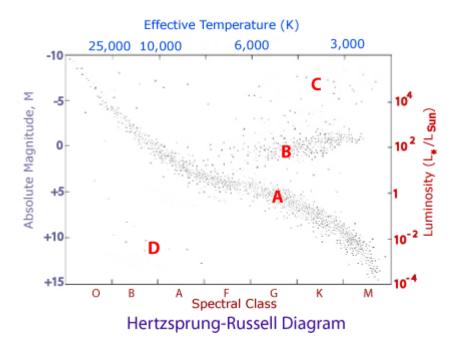


- 1. Name the type of wave that has the highest frequency.
- 2. Name the type of wave labeled C.
- 3. Name the type of wave that has the greatest energy.
- 4. Which letter shows the type of wave that can be seen by the human eye?
- 5. Name the type of wave labeled A.
- 6. Which letter indicates X-rays?
- 7. Compared to wavelengths of visible light, the wavelength of UV light is (longer/shorter/the same).
- 8. What is the relationship between frequency and wavelength? Wavelength and energy? Energy and frequency?

## Part 2 - Photoelectric Effect (Particle Behavior of Light)



- 1. What does the model about tell us in terms of frequency, energy, photons, and electrons?
- 2. Compare what is happening between the red and blue waves.
- 3. The photoelectric effect is proof that light act like what?
- 4. Photons are extremely important in the photoelectric effect. What is a photon? Why are they important?
- 5. We studied photons when we talked about the layers of stars. What can you tell me about the layer(s) of stars in terms of photons?
- 6. Are all photons able to eject electrons? Explain your answer.
- 7. Do all substances eject electron or just metals? Explain your answer.



- 1. What happens to the stars' size as you move from left to right the HR diagram?
- 2. What happens to the stars' temperature as you go from left to right on the HR diagram?
- 3. Temperature tells us the color of the stars. What color would the really hot stars be? What color would the cold stars be?
- 4. What could you tell me about the stars in region C? Region A? or Region D? (Hint: size, color, temperature, brightness)
- 5. Which group of stars is considered to have lower than average luminosity as well as higher than average temperature?
- 6. Which group of stars are near the end of their lifecycle, very large, but very cold?
- 7. What is the luminosity of The Sun?
- 8. What is the difference between apparent magnitude and absolute magnitude? What does this mean in terms of The Sun?

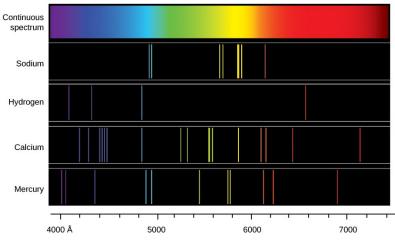
## Part 4 - Wave Behavior of Light

- 1. What two pieces of evidence do we have that tells us that light travels as waves?
- 2. Destructive interference occurs when the crest of one wave lines up with the \_\_\_\_\_ of another wave. This results in wave cancellation.
- 3. In constructive interference, the two waves are \_\_\_\_\_\_ to create a greater amplitude.
- 4. What is light diffraction?
- 5. Draw a wave. Label the crest, trough, wavelength.
- 6. Make another way with a shorter wavelength. List what is directly proportional. List what is inversely proportional.
- Make another way with a longer wavelength that the 1<sup>st</sup>. List what is directly proportional. List what is inversely proportional.

## Part 5 - Stars and Big Bang

- 1. Why do the position of stars in the universe change?
- 2. What evidence do we have that shows us this change?
- 3. What stage of life is The Sun in? What is the percentage of H and He in The Sun? Will The Sun go Supernova, why or why not?
- 4. The color a star is determined by what?
- 5. The universe was formed \_\_\_\_\_\_ billion years ago. The Sun was born \_\_\_\_\_\_ billion years ago.
- 6. Tell me what you know about The Sun? (Age, Size, Lifecycle, Color, Temperature, Brightness, etc.)
- 7. Is The Sun the brightest star in the sky? How do you know?

## Part 6 - Light Spectra



This is a bright line spectra. It shows us the difference between the continuous spectrum versus the bright line spectra of individual elements.

- 1. Explain why each element gives off different spectral lines. (Hint: excited electrons, ground state electrons, etc.)
- 2. Why would sodium give off the same color spectral lines each and every time it is heated?
- 3. When a salt or metal is heated in a flame, the flame has a distinctive color. This information was eventually extended to the study of stars. What can the color spectra of stars tell us?
- 4. Draw a model that explains how an atom gives of light/color. Be as detailed as you can. Include electrons, ground state, excited state, absorbed energy, released energy, and a bright line spectra.