## **Copper Odyssey Conversion IV**

**<u>Conversion IV</u>** - Changing copper (II) oxide to copper (II) chloride Pre-lab: Balance the following reaction.

 $\underline{\qquad} CuO_{(s)} + \underline{\ }^2_{-} HCl_{(aq)} \rightarrow \underline{\qquad} CuCl_{2(aq)} + \underline{\qquad} H_2O_{(l)}$ 

In a complete sentence describe and name the above compounds. Solid copper (II) oxide reacts with aqueous hydrochloric acid and produces aqueous copper (II) chloride and liquid water

**Observations :** Conversion III product description: **Black or Blue solid in filter paper** 

Conversion IV reaction observation: **Turns into a greenish liquid** 

## **Conversion IV Questions**

1. What type of chemical reaction is Conversion IV? <u>double displacement</u>

2. What were we trying to wash away from the black copper (II) oxide before beginning? **Any excess NaOH** 

3. Is water polar or nonpolar? \_\_\_\_\_How do you know? \_\_\_\_\_its bent shape\_\_\_\_\_\_

4. How can you tell that Conversion IV is underway? The reaction started to bubble

5. How can one tell when Conversion IV is completed? **The liquid in the flask is green** 

6. List all of the substances in the bottle at the end of Conversion IV.  $CuCl_2 \ H_2O$ 

7. Is HCl acidic or basic? \_\_\_acid\_\_\_ How do you know? \_\_\_formula starts with an H
8. Determine the number of moles of HCl in 20 mL of 6M HCl.
(Hint: convert mL to L then use Molarity=number of moles/L)
Molarity=number of moles/L
6M= x/.02 L
Moles= 0.12

11. Complete the table listing the symbol, name, classify as metal or nonmetal, give specific group, and give the symbol of another element which could be expected to react similar to the given element.

symbol	name	metal / transition metal /nonmetal	ion	Group	symbol of another element which could be expected to react similar
Cr	chromium	metal/transition	Cr <sup>2+</sup> Cr <sup>3+</sup>	Transition metal	Mo, W, same column
Fe	iron	metal/transition	+	Transition Metal	Co, Ni, Cu
Со	cobalt	metal/transition	+	<b>Transition Metal</b>	Ni, Cu, Fe
Ni	nickel	metal/transition	+	<b>Transition Metal</b>	Co, Cu, Fe
Cu	copper	metal/transition	+	<b>Transition Metal</b>	Co, Ni, Fe
Zn	zinc	metal/transition	+	<b>Transition Metal</b>	Cu, Co, Fe, Ni
As	arsenic	Nonmetal	As <sup>3-</sup>	Metalloid	Sb, P, N
Br	bromine	Nonmetal	Br-1	Halogen	F, Cl, I