Per

## **Comparing Phase Diagrams: Carbon Dioxide and Water**



## The Phase Diagrams of H<sub>2</sub>O and CO<sub>2</sub>



1. Dry ice is CO<sub>2</sub> in SOLID form. Why does dry ice create a "fog" when placed on a table at room temperature? (Hint: What phase change is occurring?)

## Looking at the phase diagrams of both carbon dioxide and water answer the following questions

What is the state of matter of $H_2O$ and $CO_2$ at point E? $H_2O$ :	CO <sub>2</sub> :	
What happens to H <sub>2</sub> O and as it moves from point E to point A?		
What happens to CO, as it moves from point E to point A?		
What is/are the state(s) of matter at point A? H <sub>2</sub> O:	CO <sub>2</sub> :	
What happens to H <sub>2</sub> O as it moves from point A to point G?		
What happens to CO <sub>2</sub> as it moves from point A to point G?		
What is/are the state(s) of matter at point G? H <sub>2</sub> O:	CO <sub>2</sub> :	
What happens to H <sub>2</sub> O as it moves from point A to point B?		
What happens to CO, as it moves from point A to point B?		
What is/are the state(s) of matter at point B? H <sub>2</sub> O:	CO <sub>2</sub> :	
What happens to H <sub>2</sub> O as it moves from point B to point H?		
What happens to CO <sub>2</sub> as it moves from point B to point H?		
What is/are the state(s) of matter at point H? H <sub>2</sub> O:	CO <sub>2</sub> :	
What happens to $H_2O$ as it moves from point B to point C?		
What happens to $CO_{2}$ as it moves from point B to point C?		
What is/are the state(s) of matter at point C? H <sub>2</sub> O:	CO <sub>2</sub> :	

## Using your answer from above:

List 3 ways the Phase Diagram for  $H_2O$  is similar to the Phase Diagram for  $CO_2$ :

List 3 ways the Phase Diagram for H<sub>2</sub>O is <u>different than</u> the Phase Diagram for CO<sub>2</sub>: